



AMS measurements of ^{55}Fe in steel

- an example of a simple analysis with a big machine

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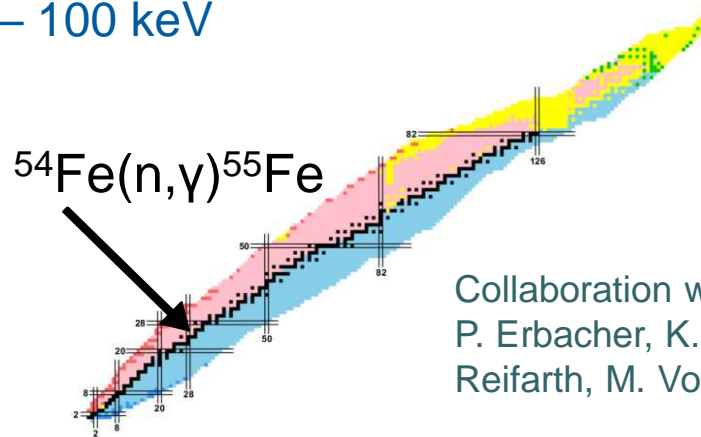
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DREAMS @ HZDR



Why ^{55}Fe ?

1) Understanding stellar nucleosynthesis requires experimental cross sections at s-process-neutron energies: $E_n \sim 1 - 100 \text{ keV}$



^{55}Co 17.54 h	^{56}Co 77.236 d	^{57}Co 271.80 d	^{58}Co 9.1h 70.8d
^{54}Fe 5.845 a	^{55}Fe 2.73 a (n, γ)	^{56}Fe 91.754	^{57}Fe 2.119
^{53}Mn $3.7 \cdot 10^6 \text{ a}$	^{54}Mn 312.2 d	^{55}Mn 100	^{56}Mn 2.5789 h

Collaboration with Z. Slavkovská, B. Brückner, P. Erbacher, K. Al-Khasawneh, S. Pavetich, R. Reifarh, M. Volknandt, M. Weigand

2) Production inside a nuclear reactor via $^{54}\text{Fe}(n, \gamma)^{55}\text{Fe}$ and $^{56}\text{Fe}(n, 2n)^{55}\text{Fe}$ on steel components (reactor vessels, tubes, steel in concrete)
“short” half-life $T_{1/2} = 2.7563 \pm 0.0004 \text{ years}$ (Pommé et al., 2019)
→ high levels in first years after shutdown

Important & hard to measure radionuclides

Cited from IAEA Radiological Characterization of Shut Down Nuclear Reactors for

Decommissioning Purposes, Technical Reports Series No. 389, Vienna 1998



TECHNICAL REPORTS SERIES NO. 389

Radiological Characterization
of Shut Down Nuclear Reactors
for Decommissioning Purposes



INTERNATIONAL ATOMIC ENERGY AGENCY, VIENNA, 1998

Fe 55
2.73 a

ϵ : no γ
 σ : 13
 $\sigma_{n, \alpha}$: 0.01

“For GCRs and other reactors, ^{55}Fe is the major short term component of the radioactive inventory following shutdown.

^{55}Fe is a hard-to-measure radionuclide...”

Ni 59
 $7.5 \cdot 10^4$ a

ϵ : β^+ ...
no γ ; σ : 77.7
 $\sigma_{n, \alpha}$: 14; $\sigma_{n, p}$: 2
 σ_{abs} : 92

“ ^{59}Ni is considered an important radionuclide for waste disposal. This is a hard-to-measure radionuclide ...”

Ca 41
 $1.03 \cdot 10^5$ a

ϵ : no γ
 σ : ~4
 $\sigma_{n, \alpha}$: 0.18
 $\sigma_{n, p}$: 0.007

“ ^{41}Ca can be of great importance when the safety of final disposal of decommissioning waste is assessed.”

Cl 36
 $3.0 \cdot 10^5$ a

σ : <10
 $\sigma_{n, \alpha}$: 0.00059
 $\sigma_{n, p}$: 0.046

“ ^{36}Cl ...is important from the viewpoint of disposal, because of its long half-life, the solubility of chloride salts, low retardation in the geosphere and potential pathways to humans.”

Zr 93
 $1.5 \cdot 10^6$ a

β^- : 0.06...
m:
 σ : <4

“ ^{93}Zr is also considered one of the critical radionuclides for long term disposal.”

in 1998: only counting techniques considered for these radionuclides

→ since 1998 AMS technique became cheaper & more common

Samples used in the study

(Merchel et al., in review at J. Radioanalytical & Nuclear Chemistry)

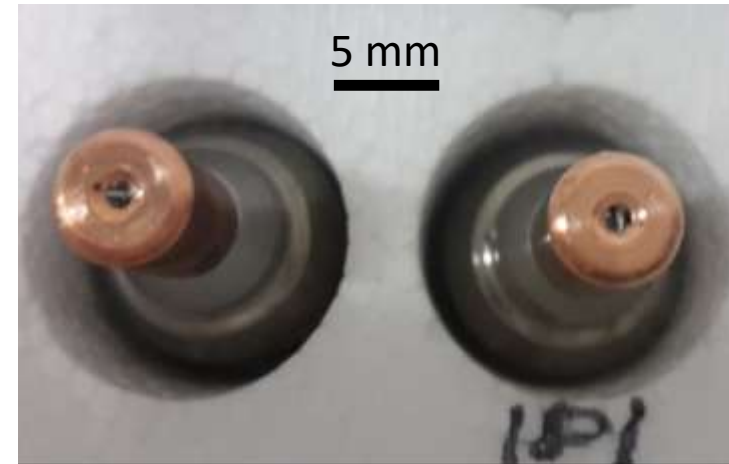
- 2 fine and 2 coarse grained steel chips drilled or rasped from a nuclear reactor vessel
- ICP-MS: measurement: steel samples can be considered 100 % Fe
→ simpler transformation of measured $^{55}\text{Fe}/\text{Fe}$ AMS ratios to specific activities

alternative preparations for each sample:

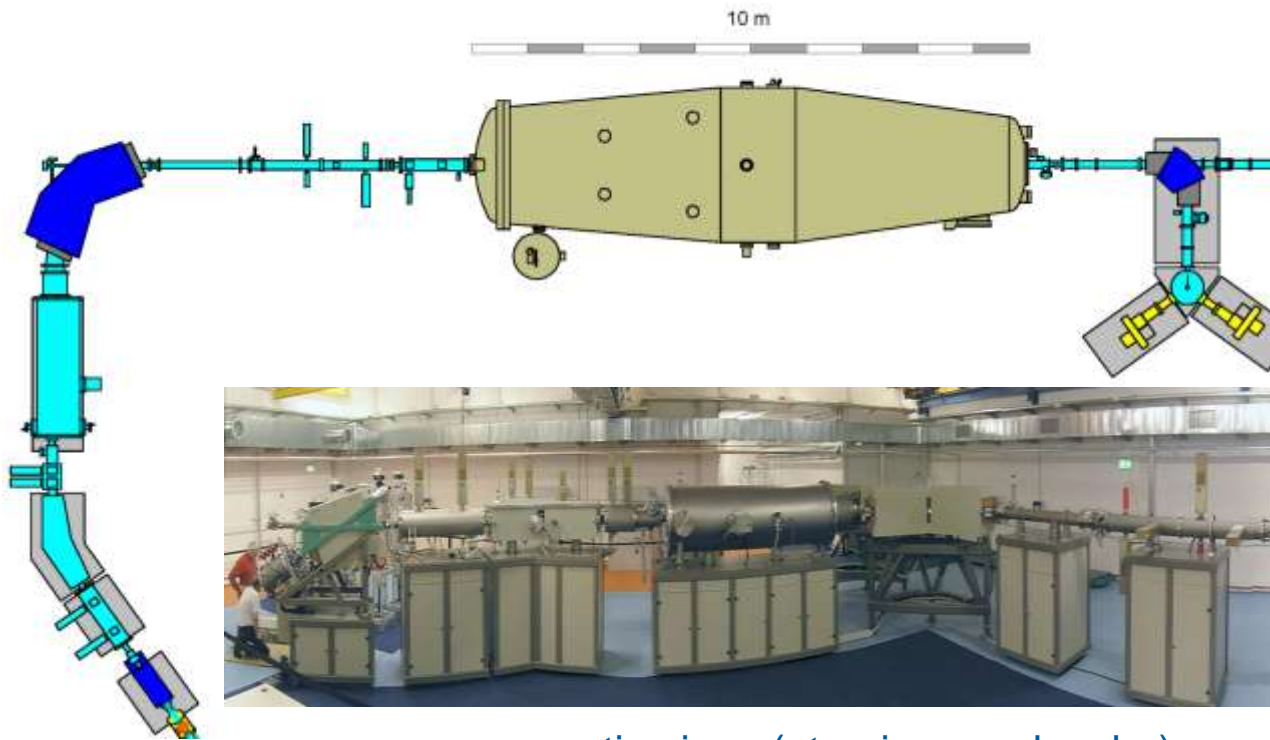
- a. steel chips (mg amounts) directly hammered in 3 “targets”
 - b. radiochemical separation on bigger solid aliquots (~1.5g): hydroxide precipitations and extraction chromatography or liquid-liquid extraction
 - 96 % used for LSC analysis
 - 4 % for Fe_2O_3 AMS target
- “5” results per sample



^{55}Fe “hard-to-measure”? – in steel it’s an easy preparation for AMS !

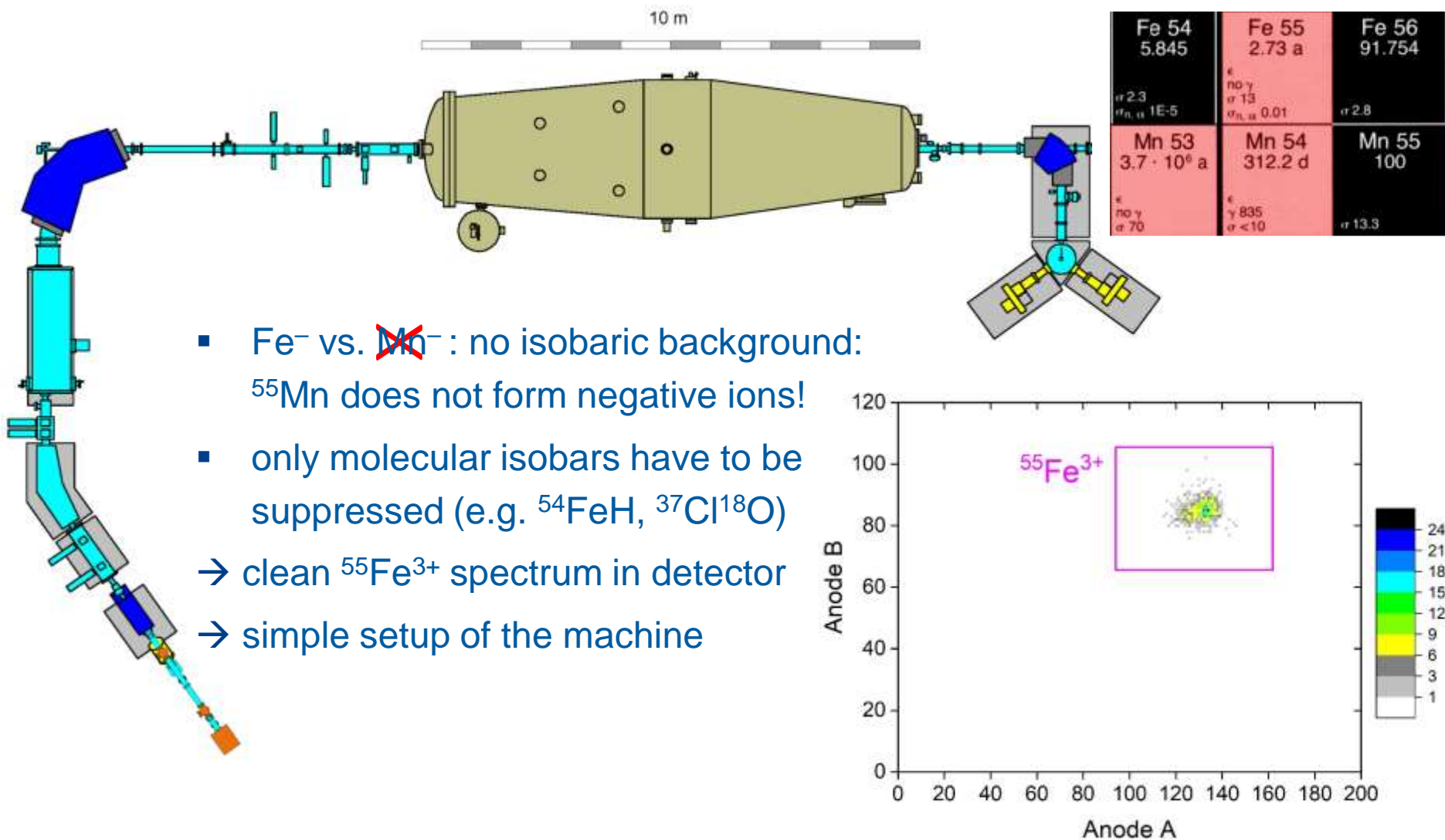


Introduction to DREsden AMS (DREAMS)



- negative ions (atomic or molecular) are extracted in Cs sputter ion source
- 1. mass separation of negative ions: Fe^-
- acceleration (voltage up to 6MV) & destruction of molecular background: stripping to positive charge states
- 2. mass separation of positive ions: Fe^{3+}
- measurement of ion currents (abundant isotopes) and detection of single ions in gas ionization detector (rare isotopes)
→ ratios $^{55}\text{Fe}/^{54}\text{Fe}$ and/or $^{55}\text{Fe}/^{56}\text{Fe}$

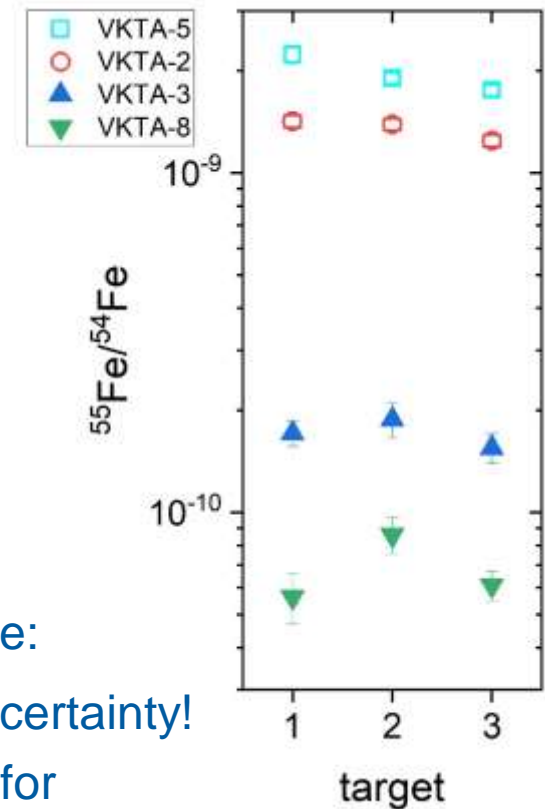
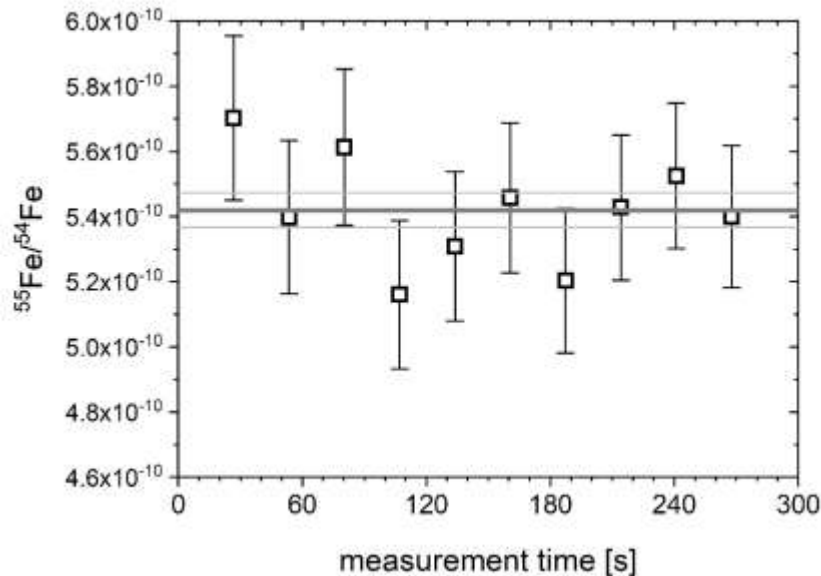
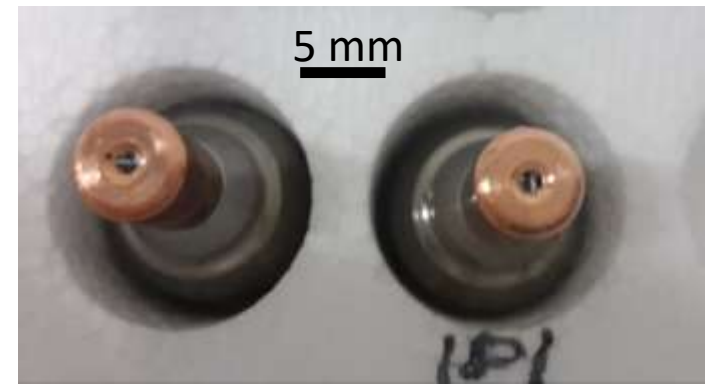
^{55}Fe “hard-to-measure”? – an easy measurement for AMS !



- Fe^- vs. ~~Mn^-~~ : no isobaric background:
 ^{55}Mn does not form negative ions!
- only molecular isobars have to be suppressed (e.g. ^{54}FeH , $^{37}\text{Cl}^{18}\text{O}$)
→ clean $^{55}\text{Fe}^{3+}$ spectrum in detector
→ simple setup of the machine

AMS-performance of steel samples

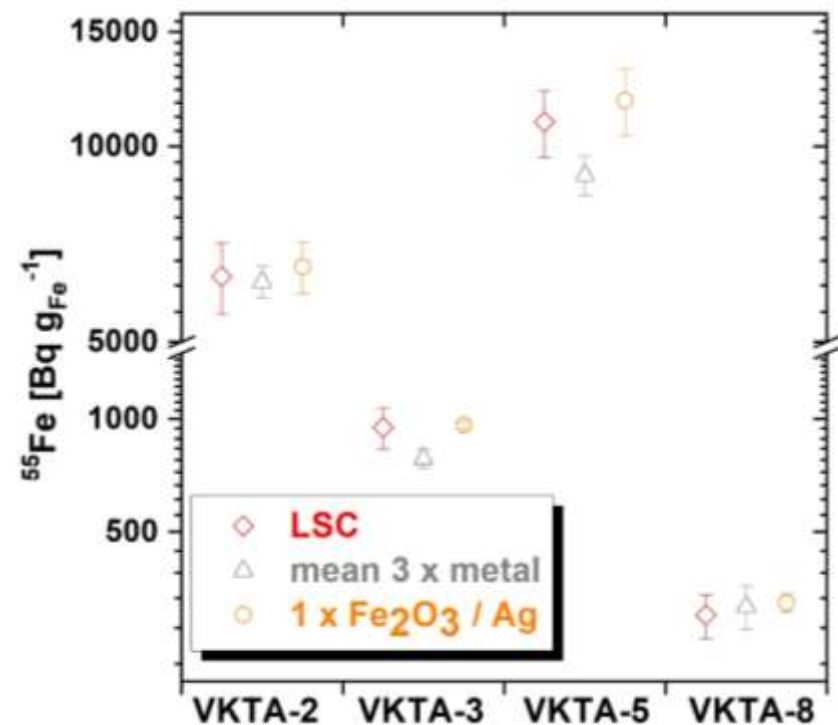
stability of a single measurement
→ overall uncertainty dominated by counting statistics!



measurements of 3 targets for each steel sample:
→ scatter between targets dominates overall uncertainty!
→ inhomogeneous steel flakes are problematic for high-precision measurements

AMS-LSC comparison

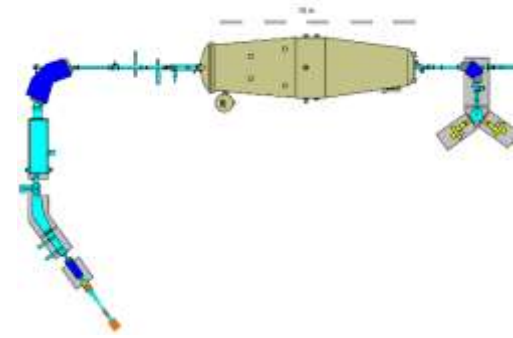
- linearity over > 2 orders of magnitude documented
- uncertainties of AMS competitive to LSC
- AMS background at single Bq / g_{Fe}



Merchel et al.,
J. Radioanalytical &
Nuclear Chemistry,
in review

	AMS	LSC
time consumption for preparation (batch of 8 steel samples)	1 hour	20 hours over ca. 4 days
measurement time per sample	15 mins	150 mins
measurement preparation time (e.g. sample loading + tuning)	6 hours	<15 mins

Funding by Helmholtz Association and European Union



- HZDR ion beam centre (IBC) offers access to its experimental facilities free of charge

→ submit your project via HZDR “GATE”:
scientific case + experimental plan
evaluation by external board

→ <https://gate.hzdr.de/user/>

HELMHOLTZ
RESEARCH FOR GRAND CHALLENGES

- EU funds Trans-National-Access via HORIZON 2020 project RADIATE relevant for users working in EU member states (except Germany) and associated states

→ submit your project for evaluation by external advisory board
→ covers cost for beam time, travel & accommodation until 12/2022

→ <https://www.ionbeamcenters.eu/radiate/radiate-transnational-access/>

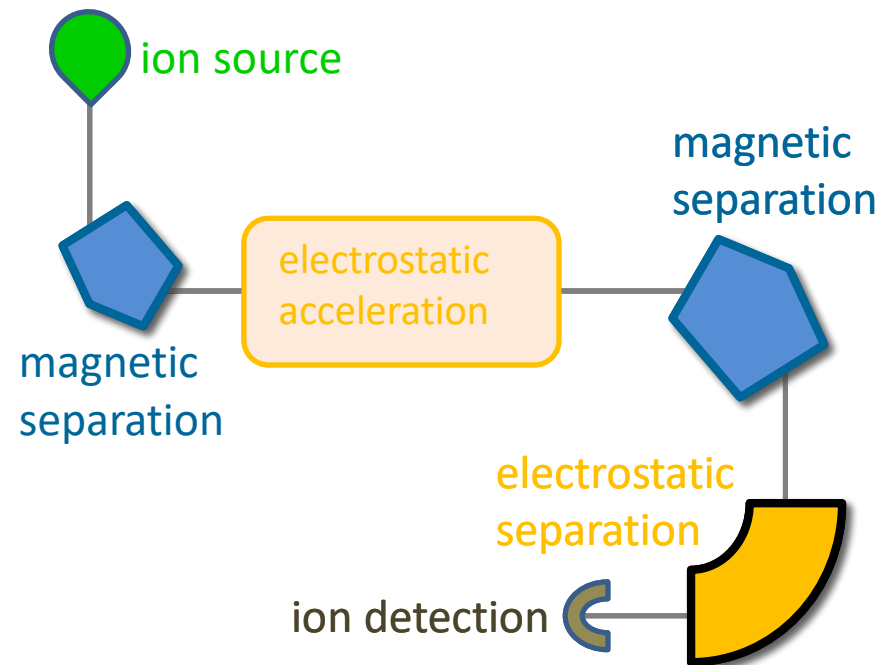
RADIATE

→ please contact j.lachner@hzdr.de

New AMS facility planned at HZDR: Start in 2023

Key features:

- smaller than DREAMS: accelerator max. 1MV
- dedicated AMS facility:
easier access & faster sample turnaround
- focus on actinides and new AMS isotopes



Beyond present standards of AMS: ILIAMS – development of AMS towards analysis of ^{59}Ni , ^{93}Zr , etc.

- deceleration step of ions before injection into accelerator
- requires high-voltage platform on ion source potential
- slow ions have time for
 - physical interaction
 - chemical transformation

Collaboration with
M. Martschini, O. Marchhart,
A. Wieser, P. Steier, R. Golser



<div>Al 26 6.35 s $1.16 \cdot 10^4$ a $\alpha = 0.20$</div>	<div>Al 27 100</div>
<div>Mg 25 10.00</div>	<div>Mg 26 11.01 $\alpha = 0.036$</div>

<div>Ni 59 $7.5 \cdot 10^4$ a $\alpha = 2.0$</div>	<div>Ni 60 26.2231</div>
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<div>Co 58 6.94 h $\alpha = 0.20$</div>	<div>Co 59 100 $\alpha = 0.20$</div>
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<div>Ru 99 12.76 $\alpha = 4$</div>	<div>Ru 100 12.60 $\alpha = 5.8$</div>	<div>Ru 101 17.06 $\alpha = 5$</div>
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<div>Tc 98 $4.2 \cdot 10^4$ a $\alpha = 0.4$</div>	<div>Tc 99 6.0 h $\alpha = 0.001$</div>	<div>Tc 100 15.8 s $\alpha = 0.001$</div>
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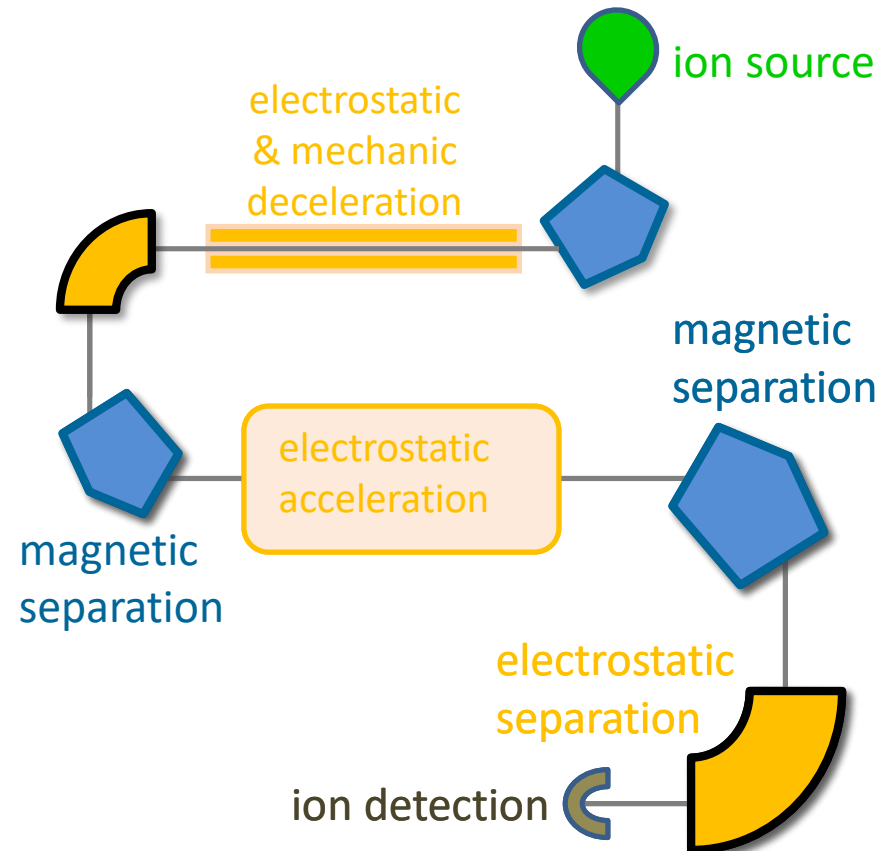
<div>Mo 97 9.56 $\alpha = 2.5$</div>	<div>Mo 98 24.19 $\alpha = 0.14$</div>	<div>Mo 99 66.0 h $\alpha = 0.14$</div>
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<div>Cl 35 75.76 $\alpha = 0.43$</div>	<div>Cl 36 $3.0 \cdot 10^5$ a $\alpha = 0.00059$</div>	<div>Cl 37 24.24 $\alpha = 0.43$</div>
<div>S 34 4.25 $\alpha = 0.25$</div>	<div>S 35 87.5 d $\alpha = 0.2$</div>	<div>S 36 0.01 $\alpha = 0.24$</div>

<div>Zr 90 51.45 $\alpha = 0.014$</div>	<div>Zr 91 11.22 $\alpha = 1.2$</div>	<div>Zr 92 17.15 $\alpha = 0.2$</div>
<div>Y 89 16.0 s $\alpha = 0.001$</div>	<div>Y 90 64.1 h $\alpha = 0.001$</div>	<div>Y 91 58.5 d $\alpha = 0.001$</div>

<div>Nb 93 16.12 s $\alpha = 0.001$</div>	<div>Nb 94 6.26 m $2 \cdot 10^4$ a $\alpha = 0.001$</div>
<div>Zr 92 17.15 $\alpha = 0.2$</div>	<div>Zr 93 $1.5 \cdot 10^4$ a $\alpha = 0.001$</div>

<div>Ba 135 26.7 h $\alpha = 0.010$</div>	<div>Ba 136 7.854 $\alpha = 0.010$</div>	<div>Ba 137 2.55 m $\alpha = 0.010$</div>	<div>Ba 138 71.696 $\alpha = 0.41$</div>
<div>Cs 134 2.90 h $\alpha = 0.010$</div>	<div>Cs 135 33 m $\alpha = 0.010$</div>	<div>Cs 136 13.16 d $\alpha = 0.010$</div>	<div>Cs 137 30.17 a $\alpha = 0.010$</div>



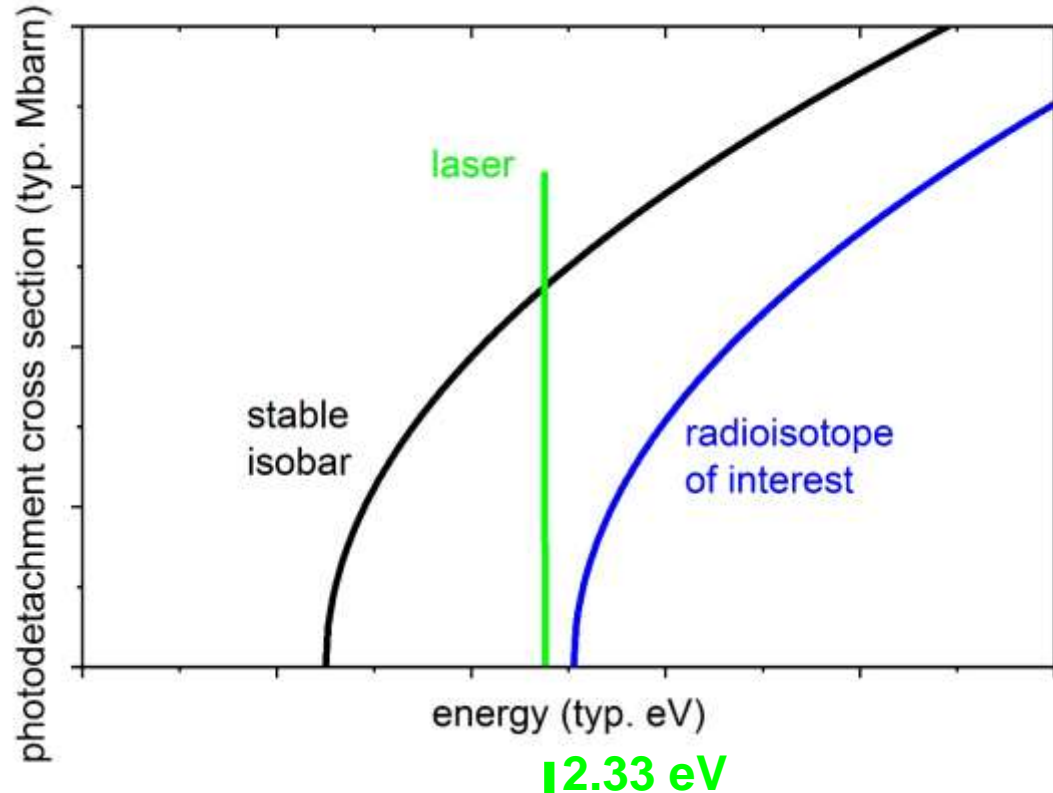
ILIAMS – development of AMS

$$N_{\text{surv}} = N_0 \cdot e^{-\sigma\phi t};$$

σ : detachment cross section,
 ϕ : photon flux,
 t : interaction time

green laser: 18 W cw
 532 nm: $E_{\text{ph}} = 2.33 \text{ eV}$

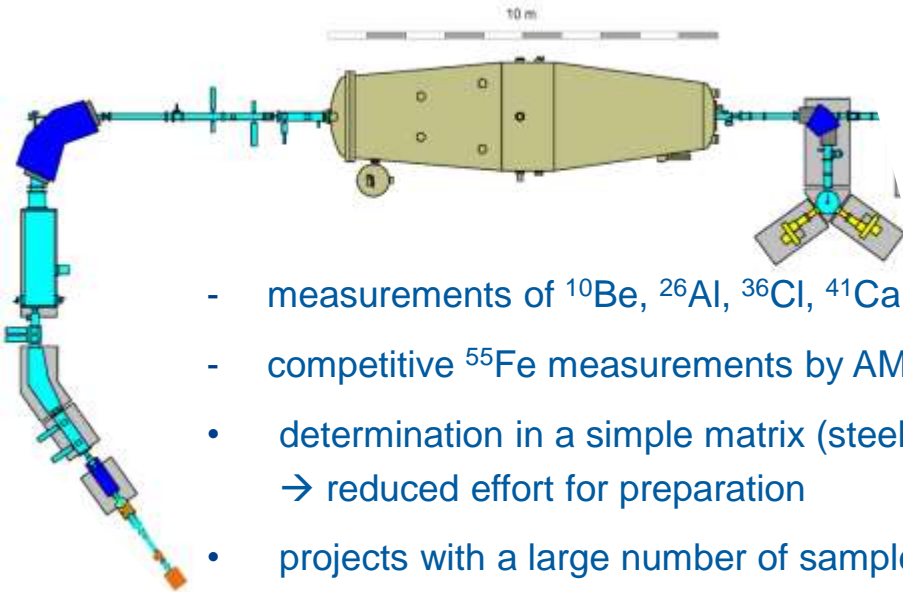
- ✓ S^- vs. Cl^-
- ✓ MgO^- vs. AlO^-
- ✓ BaF_2^- vs. CsF_2^-
- ✓ ZrF_3^- & YF_3^- vs. SrF_3^-



m=36	S^- : 2.0771029(10) eV	Cl^- : 3.612724(27) eV
m=43	MgO^- : 1.630(25) eV	AlO^- : 2.600(10) eV
m=90	ZrF_3^- , YF_3^- : < 2.3 eV	SrF_3^- : > 2.3 eV
m=173,175	BaF_2^- : < 2.3 eV	CsF_2^- : > 2.3 eV

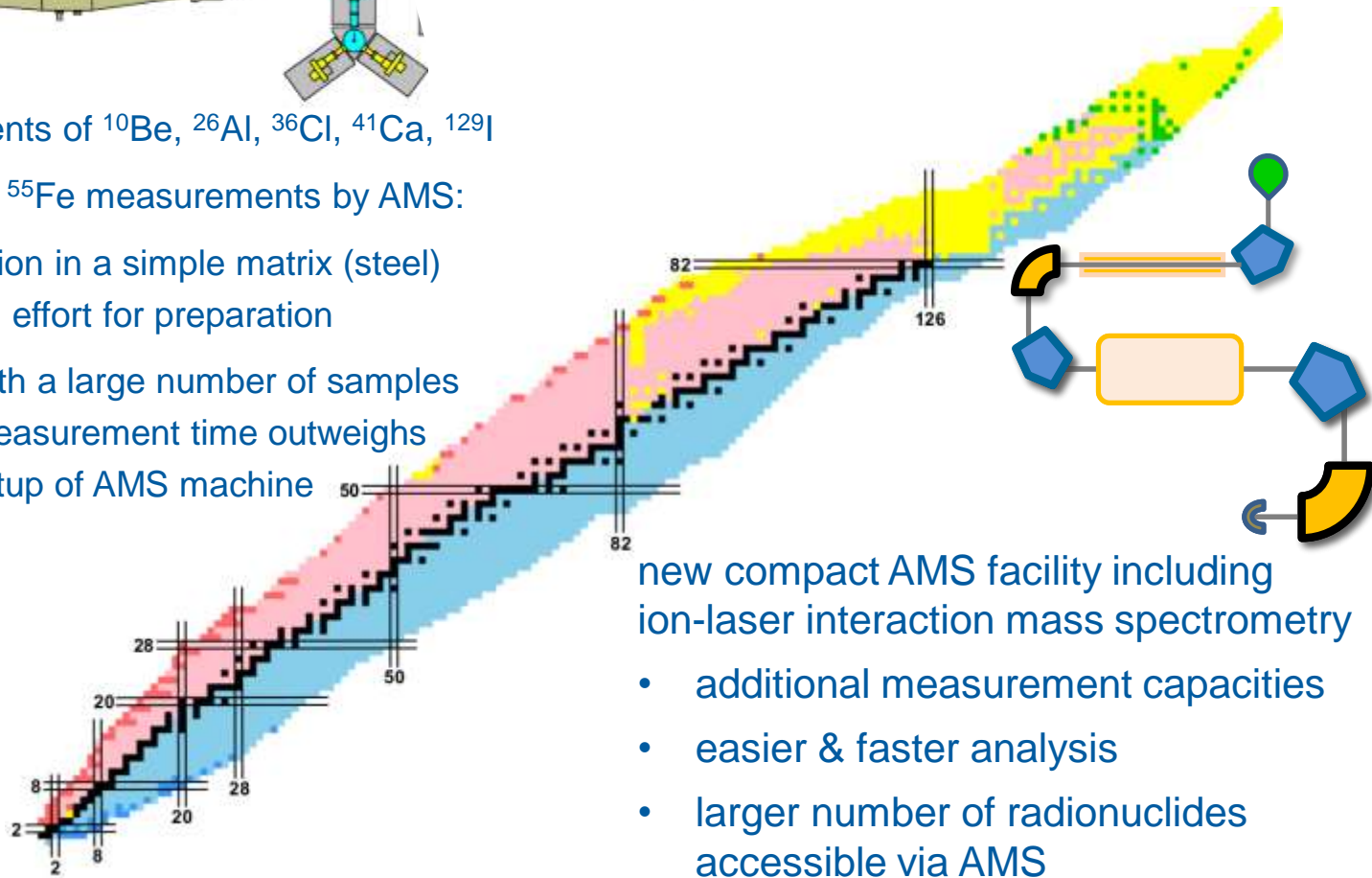
→ ^{36}Cl : Lachner et al., NIMB 2019, ^{26}Al : Lachner et al. IJMS 2021

AMS at HZDR: standard AMS radionuclides, ^{55}Fe and beyond



DREAMS @ HZDR

- measurements of ^{10}Be , ^{26}Al , ^{36}Cl , ^{41}Ca , ^{129}I
- competitive ^{55}Fe measurements by AMS:
 - determination in a simple matrix (steel)
→ reduced effort for preparation
 - projects with a large number of samples
→ short measurement time outweighs time for setup of AMS machine



new compact AMS facility including ion-laser interaction mass spectrometry

- additional measurement capacities
- easier & faster analysis
- larger number of radionuclides accessible via AMS

check for more details: www.hzdr.de/fwir